

Home Search Collections Journals About Contact us My IOPscience

Luminescence of $^{\mathbf{Cr}^{3+}}$ in lithium niobate: influence of the chromium concentration and crystal composition

This article has been downloaded from IOPscience. Please scroll down to see the full text article.

1998 J. Phys.: Condens. Matter 10 1137

(http://iopscience.iop.org/0953-8984/10/5/021)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 171.66.16.209 The article was downloaded on 14/05/2010 at 12:11

Please note that terms and conditions apply.

Luminescence of Cr^{3+} in lithium niobate: influence of the chromium concentration and crystal composition

F Lhomme†||, P Bourson†, M D Fontana†, G Malovichko†‡¶, M Aillerie† and E Kokanyan \S

† Laboratoire Matériaux Optiques à Propriétés Spécifiques, CLOES, University of Metz and Supelec, 2 rue Belin, 57078 Metz Cédex 3, France

‡ Institute for Problems of Material Sciences, Kiev, Ukraine

§ Institute of Physical Researches, Ashtarak 2, Armenia

Received 30 May 1997, in final form 26 September 1997

Abstract. The luminescence spectrum of Cr^{3+} in LiNbO₃ has been investigated, especially in the R-line range from 7200 up to 7500 Å. For this, we have studied several chromium-doped LiNbO₃ crystals with various Li/Nb ratios and different doping concentrations. In addition to the lines which were previously reported, several new peaks were detected in our investigations. In fact, the structure, the shape and the relative intensity of each line are shown to be strongly dependent on the content of doping and on the intrinsic defects related to the non-stoichiometry of the crystal. Among all peaks, we focus our attention on four main lines, which can be separated into two groups, according to their behaviours. The contribution of the first group to the whole spectrum diminishes with increasing Li/Nb ratio so that it vanishes for the stoichiometric composition. The second group of lines, occurring at higher wavelengths, is reported for the first time and is shown to be closely related to the Li/Nb ratio, especially when approaching the stoichiometric composition. These lines exhibit a similar behaviour with increasing Li/Nb ratio or Cr concentration. An explanation of the possible origins of these four peaks is given.

1. Introduction

Lithium niobate (LN) is a good optoelectronic material owing to its electro-optical and non-linear optical properties. Various applications can be quoted: optical waveguides [1] and photorefractive devices [2] as well as self-frequency doubled and self-Q-switched solid state lasers [3].

Investigations and applications of LN generally use single crystals, which are grown from a congruent composition (about 48.4 mol% Li_2O in the melt and in the crystal) and thus contain a large number of intrinsic defects associated with the deviation from the stoichiometric composition (50 mol% Li_2O in the crystal).

An appropriate doping is able to enhance interesting properties or to provide new ones. For instance, Qiu [4] proposed Cr^{3+} -doped LN as a candidate for a room-temperature tunable solid state laser owing to its strong broad-band luminescence spanning the 770–1170 nm range.

With the help of electron paramagnetic resonance (EPR) a remarkable difference of Cr^{3+} centres in LN crystals has been found [5, 6] with variation of compositions (or Li/Nb ratio) for a wide composition range (from below congruent up to stoichiometric composition).

0953-8984/98/051137+10\$19.50 © 1998 IOP Publishing Ltd

^{||} E-mail address: lhomme@ese-metz.fr

[¶] Present address: Fachbereich Physik, Universität Osnabrück, Postfach 4469, D-49069 Osnabrück, Germany.



Figure 1. Effect of Cr concentration on R-line luminescence spectra at 10 K in congruent LN crystals ($X_c = 48.4\%$).

Recent obtaining of completely stoichiometric LN crystals with properties quite different from those in congruent crystals [7] led to a renewal of interest to this material.

The present paper is devoted to the study of the luminescence spectrum recorded in doped crystals with different crystal compositions and various amounts of Cr^{3+} and their changes owing to variation of either the crystal composition or the chromium concentration.

Indeed, most of the previous experimental investigations of the luminescence of Crdoped LN were performed in congruent crystals [8–10]. In our study, we tried intentionally to separate the effect of the composition and impurity contents on Cr^{3+} luminescence spectra to investigate the independent influence of both defect subsystems. We discuss the possible origins of the main four lines in relation to the defect structure of LN.

2. Experimental results

The crystals used in our measurements were grown by the Czochralski method with various compositions and different levels of Cr doping in the melt. The composition can be evaluated



Figure 2. Effect of Cr concentration of R-line luminescence spectra in nearly stoichiometric samples ($X_c = 49.5\%$). Inset (a) the derivative of the spectrum recorded in 0.76 at.% Cr-doped crystal. Inset (b) the same spectrum corrected by the subtraction of the broad band.

by various techniques [7]. Among them, we have chosen the measurement of the width of some Raman lines, which vary linearly with the ratio $X_c = [\text{Li}]/([\text{Li}] + [\text{Nb}])$ in the crystal [11, 12].

Experiments were done with samples which were mounted in a cryostat and cooled down to 15 K. The 5145 Å Ar^+ laser line was used to pump the absorption band for the Cr^{3+} fluorescence experiments. All measurements of emission were obtained using a Spex 1401 double-grating spectrometer and a cooled RCAC 31034 PMT. The light beam was propagating perpendicular to the *c*-axis and polarized parallel to this axis.

Figure 1 shows the luminescence spectrum characteristic of the Cr-doped congruent lithium niobate ($X_c = 48.4\%$). At the shortest wavelengths are two sharp lines (L_1 and L_2), usually called R-lines and ascribed to the transition ${}^2E \Rightarrow {}^4A_2$ [8] whereas the signal at high wavelength is the tail of a broad band, which is assigned to the transition ${}^4T_4 \Rightarrow {}^4A_2$.

These spectra, as for all spectra reported in this study, are normalized to the intensity of the line L_1 . The shapes of the luminescence spectra vary with the Cr concentration (0.06, 0.7 and 3 at.%). Compared with previous data [8–10], our results allow us to point out additional features as follows. We note the increase of the intensity of the L_2 line with increasing chromium concentration. Further the position of the peak maximum and the width for both lines exhibit an upward trend. These effects are stronger for the line L_2 .

Figure 2 displays the R-lines in LN:Cr crystals with nearly stoichiometric composition $(X_c = 49.5\%)$ for 0.03, 0.14 and 0.76 at.% Cr concentration. The behaviour of the lines L₁ and L₂ appears to be similar to that observed in the congruent sample. Compared with



Figure 3. Corrected spectra of LN crystals with different compositions ($X_c = 46.7-49.85\%$) but with the same [Cr] = 0.06 at.% in the melt. Inset, composition dependence of the peak positions.

the spectra recorded in the congruent crystal, a new line and a shoulder occur at higher wavelength than the lines L_1 and L_2 . These new features can be more clearly detected if we consider the derivative of the spectrum shown in the inset (a) of figure 2. These lines denoted L_3 and L_4 are located at 7384 Å and 7454 Å respectively. In addition, lines L_5 and L_6 lying at 7324 Å and 7336 Å are observed between L_1 and L_2 .

The lines L_3 and L_4 are partly hidden by the low-frequency tail of the broad emission band. They are therefore more visible in the corrected spectra, as deduced by the substraction of the broad band, likewise in [13]. An example of the corrected spectrum is reported in the inset (b) of figure 2, which clearly shows that the lines L_3 and L_4 are much broader than the lines L_1 and L_2 . The peak position of the four lines does not show any clear trend but their intensity clearly increases with rising Cr concentration.

Up to now, we have mainly paid attention to the study of behaviour of LN:Cr luminescence with Cr concentration. Now we turn on the dependence of emission spectra on the composition in crystals with the same chromium content. Figure 3 displays the corrected spectra for the crystals LN doped with 0.06 at.% Cr but having different compositions: $X_c = 46.7$, 48.4, 49.4 and 49.85%. The spectrum exhibits four main peaks and five additional peaks which could be seen in the derived spectrum. For the lines L₁ and L₂, the change of the composition from congruent to stoichiometric induces changes in some characteristics of the luminescence spectra. The maximum position of each peak moves towards higher wavelength (see the inset) whereas the width decreases with increasing X_c . Furthermore, the change of intensity of L₂, normalized to L₁, shown in figure 3 is not



Figure 4. Cr dependence of the relative integrated intensities of the lines L_1 and L_2 for congruent samples ($X_c = 48.4\%$). *S* is the sum over all peaks.

monotonic. The intensity of both lines L_3 and L_4 increases with increasing X_c , especially close to stoichiometry.

Now we discuss the behaviour of the peaks reported in the various spectra, especially in regard to the influence of both the composition and the Cr concentration. Although a total of nine peaks was detected, we pay attention to the characteristics of the four most intense peaks: L_1 to L_4 . First, owing to their difference in their width and position, we can distinguish the lines L_1 and L_2 on one hand and the lines L_3 and L_4 on the other hand. Second, we consider the dependence of the intensity of each peak on X_c or [Cr]. In fact, we study for each peak the ratio of the integrated intensity J to the sum of areas of all peaks. This normalization of the data allows the comparison of results between various crystals. Such a procedure is valid only if the peaks arise from the same luminescence transition for different centres.

Figure 4 displays the dependence of the integrated intensity on the chromium concentration in congruent samples. The integrated intensities of L_1 and L_2 show opposite trends: $J(L_1)$ decreases for increasing Cr content whereas $J(L_2)$ displays an upward trend. $J(L_1)$ is larger than $J(L_2)$ for the lowest concentration [Cr] = 0.06 at% and a crossover occurs for larger [Cr]. We can therefore describe the effect of chromium in congruent samples from two concentration ranges for the substitution process. The first concentration range for a low Cr doping concerns a preferential substitution by the Cr³⁺ ion of the site the gives rise to the line L_1 . The second range with a higher Cr doping gives a preferential substitution in the centre leading to the line L_2 .

Figure 5 exhibits the dependence on the Cr concentration of the lines L_1 to L_4 in the crystals with the composition $X_c = 49.5\%$. The lines L_1 and L_2 exhibit a crossover similar to the situation in the congruent crystals. Moreover, it appears that the total integrated



Figure 5. Cr dependence of the relative integrated intensities of the lines L_1-L_4 for nearly stoichiometric crystal ($X_c = 49.5\%$). S is the sum over all peaks.



Figure 6. Composition dependence of the relative integrated intensities of the lines L–L₄ for the crystals with [Cr] = 0.06 at.%. *S* is the sum over all peaks.



Figure 7. Emission spectrum of a stoichiometric sample ($X_c = 50 \pm 0.01\%$) doped with [Cr] = 0.03 at.% in the melt.

intensity of the lines L_1 and L_2 is partly transferred to the lines L_3 and L_4 when [Cr] increases.

Figure 6 displays the effect of the composition in the crystal with [Cr] = 0.06 at.%. The intensities $J(L_1)$ and $J(L_2)$ compared to each other do not show a monotonic trend. With increasing X_c , the contribution of both L_1 and L_2 to the total integrated intensity diminishes to the benefit of L_3 and L_4 . This behaviour is enhanced for the crystals with compositions close to stoichiometry ($X_c = 50\%$). It is remarkable to not that this feature is quite similar on increasing both the Cr concentration and the composition X_c (see figures 5 and 6).

According to the behaviour of the integrated intensities shown in figure 6, one can expect that, in the stoichiometric crystal, the lines L_3 and L_4 should be largely dominant in the emission spectrum. This was really observed for the crystal with a controlled stoichiometric composition ($X_c = 50 \pm 0.01\%$) and doped with 0.03 at.% of chromium. The luminescence spectrum recorded for this crystal is shown in figure 7. The lines L_1 and L_2 completely vanish so that only the lines L_3 and L_4 are detected. This spectrum confirms the behaviour displayed by the lines L_1 to L_4 in figure 6. The peaks L_1 and L_2 are predominant for low X_c whereas the intensities of the lines L_3 and L_4 are large for samples with high composition

so that the opposite behaviours of L_1-L_2 and L_3-L_4 can be directly related to the defect structure of LN.

3. Discussion

In order to tentatively explain our results and interpret the origin of the different peaks it is necessary to review the previous papers on the emission spectra on LN:Cr. It is to be noted that the lines L_1 and L_2 were mainly detected in the previous studies. One of the first investigations was conducted by Glass [8] who stressed the defect structure of LN. He concluded that Cr³⁺ impurities occupy Nb sites but not Li sites and rejected the possibility of Cr substitution for structural vacancy. Later Jia et al [9] proposed that these lines could belong to a chromium on an Li site (Cr_{Li}) and to a chromium on an Nb site (Cr_{Nb}) . As the oxygen octahedron around Li is more distorted than that around the Nb on, they assigned the most intense line (here denoted L₁) located at 7306 Å mainly to Cr_{Li} and the second line peaked at 7344 Å (here denoted L_2) to Cr_{Nb} . From the calculation of field strength carried out by Glass within a point ion model, Nicholls et al [14] asserted that the fact that the Cr_{Nb} peak is lying at shorter wavelength than the Cr_{Li} line was in contradiction with the assignment of Jia et al. More recently, Michel-Calendini [15] calculated the energies of Cr^{3+} in the Li⁺ and Nb⁵⁺ sites using a molecular orbital model and found that the emission of Cr_{Li} arises at shorter wavelength than the emission of Cr_{Nb} . To explain their results, Fisher et al [13] assumed that in stoichiometric crystals the isolated Cr_{Li} centres, responsible for the line L₂, disappear and only the lines of Cr_{Li} - Cr_{Nb} dimers are observed in both absorption an emission spectra. Based on detailed analysis of optical and radiospectroscopic data, Macfarlane et al [16] concluded that the main centre that dominates in both EPR and optical absorption spectra originates from Cr³⁺ in the undistorted Li⁺ site, but the group of R-lines observed in the luminescence spectra is related to Li⁺ sites that are perturbed by nearby Nb vacancies or antisites.

Regarding our data, we can reconsider the nature of the lines L_1 and L_2 . As they are dominant in the spectra recorded in crystals with composition far from stoiciometry, they are related to the presence of intrinsic defects. However, as their behaviours with the change of Cr concentration are not the same, they probably arise from different centres. This assumption differs from the description proposed by Jia *et al* [9], in which the lines L_1 and L_2 are closely related to each other and due to dimers.

The observation of the additional lines L_3 and L_4 in LN:Cr was not previously mentioned in the literature. In fact, in a crystal codoped with Cr and Mg, Camarillo *et al* [10] have detected a new peak at higher wavelength than the line L_2 and close to our line. The assignment to a Cr³⁺ centre perturbed by Mg²⁺ ions, assumed by the authors, cannot be obviously retained in our case. The behaviours of the lines L_3 and L_4 are quite similar in intensity, width and position with increasing composition and Cr concentration. We could therefore attribute their correlated origins to the formation of dimers $Cr_{Li}^{3+}-Cr_{Nb}^{3+}$ because corresponding lines have been observed in a crystal doped with 0.2 at.% by EPR spectroscopy [6]. From atomistic-simulation study [17] it was concluded that the formation of such self-compensated impurity complexes is energetically very favourable for the LiNbO₃ lattice. So this interpretation should not be discarded.

Various models were proposed in the literature to describe the structure of intrinsic defects, responsible for deviation from the stoichiometry in LN crystals and they could be also useful to explain the possible structure of extrinsic defects. Three sites in LN lattice are in principle available for the substitution by the dopant Cr^{3+} . They are the centres of

the oxygen octahedra which are normally occupied by the sequence of Li, Nb, empty site along the c-axis. Among these three sties, the structural vacancy was generally not invoked as a possible site substitution by Cr [8, 9].

Three simple models were proposed to describe the defect structures in LN. Fay *et al* [18] suggested a model assuming that oxygen and lithium vacancies are the main non-stoichiometric defects in LN with a corresponding formula as:

$$[Li_{1-x}V_{Li'x}][Nb][O_{3-x/2}V_{O-x/2}].$$

But this model should lead to a lower density of the crystal than derived from the density measurement [19]. Two other models were then proposed assuming the location of niobium in Li site, the so-called niobium antisites. These models differ in the description of the mechanism of the charge compensation. For Lerner *et al* [20], the charge compensation is due to lithium vacancies and corresponds to the formula:

$$[Li_{1-5x}V_{Li'4x}Nb^{4}][Nb][O_3]$$

whereas for Abrahams and Marsh [21], the charge compensation of Nb_{Li} is achieved by a niobium vacancy, leading to the following formula:

$$[\text{Li}_{1-5x}\text{Nb}^{4} \cdot 5x][\text{NbV}_{Nb4x}][\text{O}_3].$$

Since the lines L_1 and L_2 disappear in the completely stoichiometric sample, we consider that they are related to Cr ions having intrinsic defects as charge compensators in their nearest surroundings. Thus, in the frame of the two last models, L_1 and L_2 could be linked to niobium antisites (Nb_{Li}) and cationic vacancies (V_{Nb} or V_{Li}). the lines L_3 and L_4 could arise from the proper effect of Cr ions as extrinsic defects in the structure.

4. Summary

From the luminescence experiments performed on various LN:Cr crystals with different compositions and Cr concentrations, we were able to show the distinct effects of the chromium concentration and sample composition. Four main peaks called L_1 to L_4 have been pointed out in our spectra instead of two lines (L_1 and L_2) in the previous studies. Up to nine peaks have been resolved in the R-line range of our luminescence spectra. In order to understand the origin of these peaks, the behaviour of their characteristics (position, width, integrated intensity) were analysed under variation of Li/Nb ratio and/or doping concentration.

We can distinguish the main groups L_1-L_2 and L_3-L_4 . The effect of Cr doping on the lines L_1 and L_2 can be described by two substitution ranges. The first range for low Cr doping gives rise to the preferential creation of the L_1 centre whereas the second range for a higher Cr doping corresponds rather to the creation mainly of the L_2 chromium centre. In a nearly stoichiometric composition, the lines L_3 and L_4 occur and their contribution to the whole spectra, compared with that arising from the lines L_1 and L_2 , increases with rising Cr concentration.

The increase of the Li content in the crystal leads to the growth of relative intensity of lines L_3 and L_4 , similar to the influence of Cr concentration. The absence of L_1 and L_2 lines in a stoichiometric sample confirms the supposition that they are related to the complex centres due to Cr and intrinsic defects.

Acknowledgments

We are grateful to K Polgar and V Grachev for useful discussions and help in the spectrum treatment. Some of the studied crystals were kindly given by V P Yarunichev and V T Gabrielyan.

References

- [1] Armenise M N, Canali C, De Sario M and Zanoni E 1993 Mater. Chem. Phys. 9 267
- [2] Krätzig E and Schirmer O F 1988 Photorefractive Materials and their Applications I (Berlin: Springer)
- [3] He Q, De Michelli M P, Ostrowsky D B, Lallier E, Pocholle J P, Papuchon M, Armani F, Delecourt D, Grezes-Besset G and Pelletier E 1992 Opt. Commun. 89 54
- [4] Qiu Y 1993 J. Phys.: Condens. Matter 5 2041
- [5] Malovichko G I, Grachev V G, Kokanyan E P and Gabrielyan V T 1986 Sov. Phys.-Solid State 28 1453
- [6] Grachev V, Malovichko G and Schirmer O 1996 Ferroelectrics 185 5
- [7] Malovichko G, Grachev V, Kokanyan E P, Schirmer O F, Betzler K, Gather B, Jermann F, Klauer S, Schlarb U and Wöhlecke M 1993 Appl. Phys. A 56 103
- [8] Glass A M 1969 J. Phys. Chem. 50 10908
- [9] Jia W, Liu H, Knutson R and Yen W N 1990 Phys. Rev. B 41 10906
- [10] Camarillo E, Garcia-Solé J, Cusso F, Aguillo-Lopez F and Sanz-Garcia J A 1991 Chem. Phys. Lett. 185 505
- [11] Schlarb U, Klauer S, Wesselmann M, Betzler K and Wöhlecke M 1993 Appl. Phys. A 56 311
- [12] Ridah A, Bourson P, Fontana M D and Malovichko G 1996 J. Phys.: Condens. Matter 9 9687-93
- [13] Fisher C, Kapphan S, Feng X-Q and Cheng N 1995 Radiat. Eff. Defects Solids 133 199
- [14] Nicholls J F H, Han T P J, Henderson B and Jaqué F 1993 Chem. Phys. Lett. 202 560
- [15] Michel-Calendini F 1994 Radiat. Eff. Defects Solids 134 175
- [16] Macfarlane P I, Holliday K, Nicholls J F H and Henderson B 1995 J. Phys.: Condens. Matter 7 9643
- [17] Donnerberg H, Tomlinson S M, Catlow C R A and Schirmer O F 1989 Phys. Rev. B 40 201
- [18] Fay H, Alford W J and Dess H M 1968 Appl. Phys. Lett. 89 12
- [19] Kovacs L and Polgar K 1986 Cryst. Res. Technol. 21 6
- [20] Lerner P, Legras C and Duman P J 1968 J. Crystal Growth 3/4 231
- [21] Abrahams S C and Marsh P 1986 Acta Crystallogr. B 2 61